# **Basis set optimization**

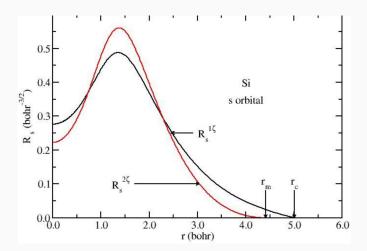
13/03/2024 - Federico Pedron

### Key concepts

siesta can automatically generate basis sets, or you can provide whatever radial function you want.

Basis functions become strictly zero beyond a certain radius, *rcut*.

For multiple-z basis, the second-z orbital is equal to the first-z orbital beyond a matching radius *rmatch*.



**Optimizing a Basis Set** 

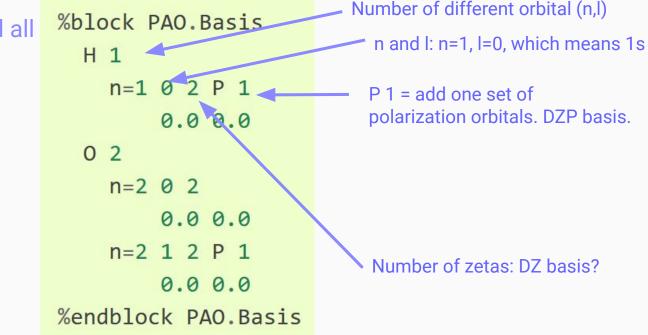
# Exploring the PAO.Basis block

We have each species and all orbitals with different (n,l) separated.

For water:

H -> 1s

0 -> 2s, 2p



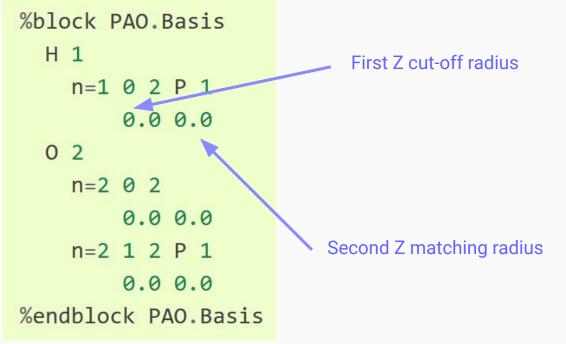
## Exploring the PAO.Basis block

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# **Basis Enthalpy**

- We want to get a good energy for a set of orbitals.
- We don't want those orbitals to get needlessly large.

Basis Enthalpy = E<sub>total</sub> + "P<sub>basis</sub> . V<sub>orbitals</sub>"

# **Basis Enthalpy**

• Not a real physical magnitude, we choose it as a input value.

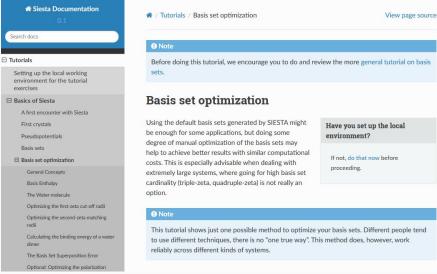
BasisPressure 0.2 GPa

• The **0.2 GPa** default works well for most cases, but for first- and second-row elements, it might result in very short orbitals. Use **0.02 GPa** instead.

#### Go to 03-BasisSets, and visit:

#### https://docs.siesta-project.org/projects/siesta/en/latest/tutorials/basic/basisoptimization/index.html

Follow the first two practical sections: **Optimizing the First-Zeta cutoff radii**, and **Optimizing the Second-Zeta matching radii**.



**Testing the Basis Set** 

# Testing the optimized basis

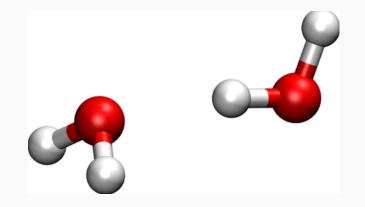
How do we know if we effectively have a better basis set than the default?

At least, three things are important to check:

- Costs
- Quality
- Transferability

We need to test this in a slightly different system!

# Binding energy of a water dimer



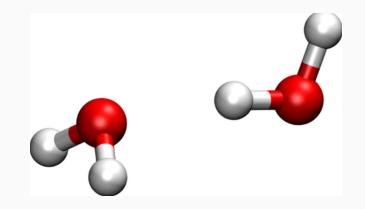
# Testing the optimized basis

Run the third part of the tutorial, *Calculating the binding energy of a water dimer*.

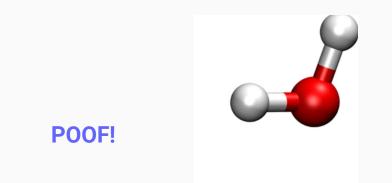
Did we get better results with our optimized basis set?

**Appendix: Ghost atoms** 

#### **Basis Set Superposition Error**



### **Basis Set Superposition Error**



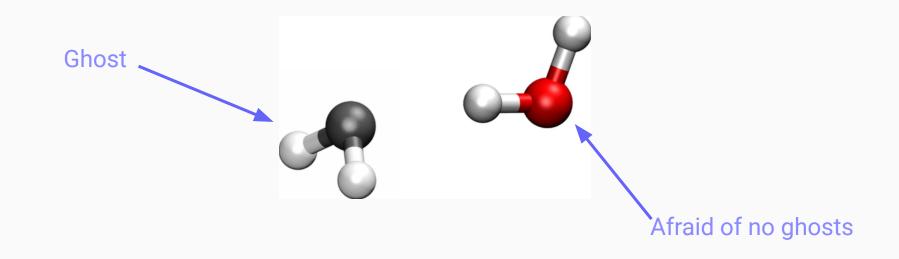
We lost the basis functions for the second molecule! What if they are important?

#### **Ghost atoms**

We add the basis functions that would belong to an atom, if the atom were there.

We do not add electrons or nuclei to the calculation!

#### **Ghost atoms**



### **Ghost atoms**

To add ghost atoms, we just create a new species with **negative atomic number**.

```
NumberOfSpecies 4
%block ChemicalSpeciesLabel
1 8 0
2 1 H
3 -8 0_ghost
4 -1 H_ghost
%endblock ChemicalSpeciesLabel
```

Yes, this means we have to duplicate the pseudopotential files and add extra terms to the PAO.Basis block.